

### DPP No. 15

Total Marks: 34

Max. Time: 40 min.

**Topic: General Organic Chemistry** 

#### Type of Questions

Single choice Objective ('-1' negative marking) Q.1 to Q.2

Multiple choice objective ('-1' negative marking) Q.3 Match the Following (no negative marking) Q.4

Subjective Questions ('-1' negative marking) Q.5 to Q.8

M.M., Min.

[6, 6] (3 marks, 3 min.)

(4 marks, 4 min.) [4, 4]

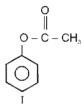
(8 marks, 10 min.) [8, 10]

(4 marks, 5 min.) [16, 20]

1. The reactivity order of benzene ring for the given reaction is (benzene ring with highest  $\pi$  electron density

will be most reactive)





- (A) III > I > IV > II
- (C) III > IV > II > I

- (B) I > III > IV > II
- (D) |V > |I| > |I|
- 2. Arrange the following compounds in the order of decreasing reactivity towards electrophilic substitution









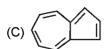


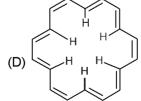
- (A) V > IV > III > II > I
- (C) I > II > IV > III > V

- (B) I > II > III > V > IV
- (D) I > III > IV > II > V
- 3.\* Which of the following are Aromatic in nature.









4. Match the following:

#### Column - I

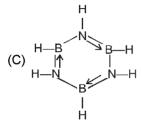
#### Column - II



(p) Aromatic



(q) Non aromatic



(r) Anti aromatic



- (s) Heterocyclic
- 5. How many of the following compounds are more reactive than benzene towards electrophilic substitution.













6. How many species out of the following are aromatic?











- 7. Explain the terms Inductive and Electromeric effects.
- 8. Which bond is more polar in the following pairs of molecules?
- (a)  $H_3C H$   $H_3C Br$  (b)  $H_3C NH_2$   $H_3C OH$  (c)  $H_3C OH$   $H_3C SH$



# swer Ke

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1.

(D)

2.

(B)

3.

(BCD)

4.

(A - p,s); (B - p,s); (C - p,s); (D - r) 5.

6.

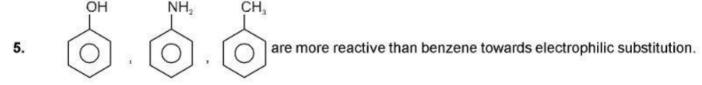
2

## nts & Solutions

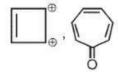
**DPP No. #15** 

1. (D) Due to +m order

- 2. Rate of electrophilic substitution reaction ∞ Stability of arenium ion.
- 4. Aromatic  $\rightarrow$  planar, cyclic, (4n+2)  $\pi e^{\Theta}$ , complete conjugation Antiaromatic  $\rightarrow$  planar, cyclic, (4n)  $\pi e^{\Theta}$ , complete conjugation Non aromatic-cyclic structure with non-planar geometry with any hybridization



Aromatic species are 6.



7. Inductive effect is permanent displacement of shared pair of electron along the chain of carbon atom due to presence of polar covalent bond.

Electromeric effect is a temporary effect. It is defined as the complete transfer of a shared pair of  $\pi$ electrons to one of the atoms joined by a multiple bond on the demand of an attacking reagent.

8.

(a) H<sub>2</sub>C - Br (b) H<sub>2</sub>C - OH (c) H<sub>2</sub>C - OH

